Copper, palladium and platinum nanostructures with controlled morphologies via polyethyleneimine assisted chemical synthesis

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Abstract

Metals with excellent optical or catalytic properties have drawn important interest for their fabrication in the nanoscale by 'bottom-up' approaches. The small size of the nanoparticles together with a nonisotropic shape can induce certain differences in the acquired material properties in comparison with the 'bulk' ones. However, metals such as Cu, Pd and Pt possess an intrinsic highly symmetric facecentered-cubic (fcc) crystal structure. Therefore, the insertion of shape-directing agents such as proper surfactants is needed to drive the particles growth to an anisotropic mode.

Polyethyleneimine (PEI) is a hydrophilic polymer with primary, secondary and tertiary amino groups and an abundant positive charge. In this work, we present the use of PEI as a stabilizer and shapemodifying agent for the tailored synthesis of water-dispersible Cu, Pd and Pt nanostructures with various morphologies. Typical shapes include for instance copper nanoparticles (Fig. 1), palladium nanodendrites (Fig. 2) and platinum nanoflowers (Fig. 3). The co-reducing role of PEI is also presented, depending also on the rest synthesis parameters used (such as the choice of solvent-reductant and the reaction temperature). We also compare the role of different reaction conditions (type and concentration of solvents, surfactants, reducing agents, precursors as well as temperatures) for the successful preparation of anisotropic shapes for the systems under study. Moreover, we present and discuss the basic physical properties (optical, electrocatalytic) of our nanomaterials.

Figures

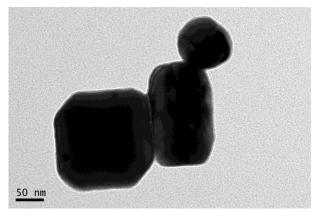


Fig. 1 TEM image of copper nanostructures prepared in the presence of PEI and diethyleneglycol

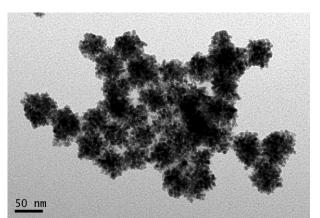


Fig. 2 Palladium nanodendrites obtained by reducing K₂PdCl₄ in the presence of oleylamine and polyethyleneimine

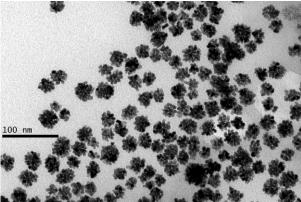


Fig. 3 Flower-like platinum nanostructures synthesized by the thermolytic reduction of platinum acetylacetonate in a DEG/PEI mixture